## University of the West of Scotland

### Module Descriptor

## Session: 2024/25

Title of Module: Properties of Matter						
Code: PHYS08004	SCQF Level: 8 (Scottish Credit and Qualifications Framework)	Credit Points: 20	ECTS: 10 (European Credit Transfer Scheme)			
School:	School of Computing, Engineering, and Physical Sciences					
Module Co-ordinator:	Marcus Scheck					
Summary of Module						
This module is a core module at Level 8 on Institute of Physics (IoP) accredited Physics programmes. The aim of this module is to develop understanding of the properties of condensed, liquid, and gaseous matter at an introductory level.						
The module covers the following	g topics.					
Distinction between solids, liquid bonds; interatomic forces/potent	ds and gases; ionic, c tial energies; fluid pre	ovalent, Van der Waa ssure; buoyancy; Ber	als, and metallic noulli's			

bonds; interatomic forces/potential energies; fluid pressure; buoyancy; Bernoulli's equation; surface tension; viscosity; elastic properties of solids; thermal expansion of solids' crystal structures; point defects, dislocations and grain boundaries.

The introductory thermodynamics aspects of the course include the Kelvin temperature scale, properties of ideal gases, equation of state of ideal and real gases, 0<sup>th</sup> and 1<sup>st</sup> law of thermodynamics, kinetic gas theory, Maxwell-Boltzmann distribution, 2<sup>nd</sup> law of thermodynamics, entropy, heat engines, Carnot cycle, microscopic interpretation of entropy, transport phenomena.

The module also includes **practical lab classes** which enable students to put into practice the principles covered in the lectures and tutorials. They cover topics on viscosity in liquids, ideal gas law, adiabatic processes, latent heat, and the elastic and hydrostatic properties of matter.

We have defined a set of Graduate Attributes that are the skills, personal qualities and understanding to be developed through your university experience that will prepare for life and work in the 21st century (https://www.uws.ac.uk/current-students/your-graduate-attributes/). The Graduate Attributes relevant to this module are listed below.

- Graduate Attributes Academic: critical thinker; analytical; inquiring; knowledgeable; digitally literate; problem solver; autonomous; incisive; innovative
- Graduate Attributes Personal: effective communicator; influential; motivated
- Graduate Attributes Professional: collaborative; research-minded; enterprising; ambitious; driven

Module Delivery Method								
Face-To- Face	Blended	Fully Online	HybridC	Hybrid 0	Work-Based Learning			
$\boxtimes$								
See Guidance Note for details.								

# Campus(es) for Module Delivery

The module will **normally** be offered on the following campuses / or by Distance/Online Learning: (Provided viable student numbers permit) (tick as appropriate)

Paisley:	Ayr:	Dumfries:	Lanarkshire:	London:	Distance/Online Learning:	Other:
$\boxtimes$						Add name

Term(s) for Module Delivery						
(Provided viable student numbers permit).						
Term 1		Term 2	$\boxtimes$	Term 3		

Learn These appro At the	Learning Outcomes: (maximum of 5 statements) These should take cognisance of the SCQF level descriptors and be at the appropriate level for the module. At the end of this module the student will be able to:						
L1	Relate the concepts of temperature and pressure to the properties of an ideal gas						
L2	State the laws of thermodynamics and apply them to simple physical systems.						
L3	Apply the basics of kinetic theory to problems involving gases.						
L4	Describe and explain the properties and behaviour of liquids and solids.						
L5	Conduct prescribed laboratory experiments, record and analyse data, estimate experimental uncertainties and draw conclusions. Record experimental procedures and observations in a lab report.						

Employability Skills and Personal Development Planning (PDP) Skills

SCQF Headings	During completion of this module, there will be an opportunity to achieve core skills in:				
Knowledge and Understanding (K and U)	<ul> <li>SCQF Level 8</li> <li>A critical understanding of underlying theories governing properties of matter.</li> <li>A critical approach towards problem solving</li> </ul>				
Practice: Applied Knowledge and Understanding	<ul> <li>SCQF Level 8</li> <li>Using a selection of mathematical skills, techniques and practices applicable to modern day physics.</li> <li>Confirmation of theoretical models with laboratory experiments.</li> <li>Practicing literature searches and experimental methodologies such as uncertainty evaluation.</li> </ul>				
Generic Cognitive skills	SCQF Level <b>8</b> Critical appreciation of analysis, evaluation a	of underlying physical concepts. Problem and solving.			
Communication, ICT and Numeracy Skills	<ul> <li>SCQF Level 8</li> <li>Use of calculators and computers.</li> <li>Literary skills, enabling the communication of obtained results e.g. lab-report.</li> </ul>				
Autonomy, Accountability and Working with others	<ul> <li>SCQF Level 8</li> <li>Individual studying and small project management.</li> <li>Working towards deadlines and avoiding unnecessary penalties.</li> <li>Team-working abilities, as lab work is encouraged to be done in groups.</li> </ul>				
Pre-requisites:	Before undertaking th undertaken the follow	his module the student should have <i>r</i> ing:			
	Module Code:           PHYS07006           PHYS07007           MATH07003           MATH07009           Other:	Module Title: Introductory Physics A Introductory Physics B Applied Mathematics Mathematical Analysis or equivalent			
Co-requisites	Module Code:	Module Title:			

\*Indicates that module descriptor is not published.

Learning and Teaching

In line with current learning and teaching principles, a 20-credit module includes 200 learning hours, normally including a minimum of 36 contact hours and maximum of 48 contact hours.

Learning Activities During completion of this module, the learning activities undertaken to achieve the module learning outcomes are stated below:	Student Learning Hours (Normally totalling 200 hours): (Note: Learning hours include both contact hours and hours spent on other learning activities)
Lecture/Core Content Delivery	24
Tutorial/Synchronous Support Activity	12
Laboratory/Practical Demonstration/Workshop	12
Independent Study	152
	200 Hours Total

## \*\*Indicative Resources: (eg. Core text, journals, internet access)

The following materials form essential underpinning for the module content and achieving the learning outcomes:

B.H Flowers and E. Mendoza: Properties of Matters [ISBN 0471 26497 0]

Material Sciences (Chapter 9): https://openstax.org/details/books/university-physics-volume-3

Fluid Mechanics (chapter 8): https://openstax.org/details/books/university-physics-volume-1

Introduction to Thermodynamics: https://openstax.org/details/books/university-physics-volume-2

(\*\*N.B. Although reading lists should include current publications, students are advised (particularly for material marked with an asterisk\*) to wait until the start of session for confirmation of the most up-to-date material)

#### **Attendance and Engagement Requirements**

In line with the <u>Student Attendance and Engagement Procedure</u>: Students are academically engaged if they are regularly attending and participating in timetabled on-campus and online teaching sessions, asynchronous online learning activities, course-related learning resources, and complete assessments and submit these on time.

Equality and Diversity

The University's Equality, Diversity and Human Rights Procedure can be accessed at the following link: <u>UWS Equality, Diversity and Human Rights Code.</u>

(N.B. Every effort will be made by the University to accommodate any equality and diversity issues brought to the attention of the School)

#### **Supplemental Information**

Divisional Programme Board	Physical Sciences
Assessment Results (Pass/Fail)	Yes □No ⊠
School Assessment Board	Physical Sciences
Moderator	Des Gibson
External Examiner	H Boston
Accreditation Details	Institute of Physics
Changes/Version Number	<b>3.0</b> Module descriptor amended to conform to the new template format and to reflect outcomes from ILR 2023.

#### Assessment: (also refer to Assessment Outcomes Grids below)

Assessment 1 - Class Test (60%)

Assessment 2 – Coursework Assignment worth 20% of the final grade

Assessment 3 – 4x laboratory experiments, in sum worth 20% of the final grade

(N.B. (i) **Assessment Outcomes Grids** for the module (one for each component) can be found below which clearly demonstrate how the learning outcomes of the module will be assessed.

(ii) An **indicative schedule** listing approximate times within the academic calendar when assessment is likely to feature will be provided within the Student Module Handbook.)

Assessment Outcome Grids (See Guidance Note)

Component 1							
Assessment Type (Footnote B.)	Learning Outcome (1)	Learning Outcome (2)	Learning Outcome (3)	Learning Outcome (4)	Learning Outcome (5)	Weighting (%) of Assessment Element	Timetabled Contact Hours

Class Test	<	~	<	~		60	2
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Component 2								
Assessment Type (Footnote B.)	Learning Outcome (1)	Learning Outcome (2)	Learning Outcome (3)	Learning Outcome (4)	Learning Outcome (5)	Weighting (%) of Assessment Element	Timetabled Contact Hours	
Course work	~	~	~	✓		20	0	

Component 3							
Assessment Type (Footnote B.)	Learning Outcome (1)	Learning Outcome (2)	Learning Outcome (3)	Learning Outcome (4)	Learning Outcome (5)	Weighting (%) of Assessment Element	Timetabled Contact Hours
Laboratory Experiments	~	<b>~</b>	✓	<	<b>~</b>	20	12
Combined Total for All Components					100	14	