## University of the West of Scotland

### Module Descriptor

### Session: 2024/25

Title of Module: Organic Chemistry 3								
Code: CHEM09004	SCQF Level: 9 (Scottish Credit and Qualifications Framework)	Credit Points: 20	ECTS: 10 (European Credit Transfer Scheme)					
School:	School of Computing, Engineering and Physical Sciences							
Module Co-ordinator:	Ciaran T Ewins							

### Summary of Module

The first section of the module will deals with the important reactions; nucleophilic substitution reactions ( $S_N1$  and  $S_N2$ ), alkene forming elimination reactions (E1 and E2 and E1<sub>cb</sub>), formation and hydrolysis of esters.

There is a study of carbanions and enolate ions and their characteristic reactions such as Adol and Claisen reactions and their importance in organic synthesis. The section on carbanions is then extended to include the synthetic importance of acetoacetic ester and diethyl malonate as well as Michael additions. The concept of the use of protecting and blocking groups in organic chemistry will then be introduced with particular application to the synthesis of peptides. Structure and properties of man-made and naturally occurring polymers including common addition and condensation polymers, carbohydrates and proteins.

Stereochemistry and synthesis of heterocycles will be introduced.

Applications of spectroscopy and computer modelling in organic chemistry will be introduced. Those who complete this module will have developed competencies in report writing, working to deadlines. There is a lab programme which includes the aldol condensation, pyrrole and coumarin synthesis formation and a green Wittig reaction. The important organic chemistry techniques of purification, chromatography and spectroscopy are used extensively in these labs.

Module Delivery Method								
Face-To- Face	Blended	Fully Online	HybridC	Hybrid 0	Work-Based Learning			
$\boxtimes$								

See Guidance Note for details.

### Campus(es) for Module Delivery

The module will **normally** be offered on the following campuses / or by Distance/Online Learning: (Provided viable student numbers permit) (tick as appropriate)

Paisley:	Ayr:	Dumfries:	Lanarkshire:	London:	Distance/Online Learning:	Other:
$\boxtimes$						Add name

Term(s) for Module Delivery								
(Provided viable student numbers permit).								
Term 1	Term 1 Image: Imag							

These appro	Learning Outcomes: (maximum of 5 statements) These should take cognisance of the SCQF level descriptors and be at the appropriate level for the module. At the end of this module the student will be able to:						
L1		ompetence in applying the key principles and theories relating to and mechanisms of organic chemical reactions					
L2	Display a critica	I understanding of the use of carbanions in organic synthesis					
L3	Display a detaile	ed knowledge of man-made and natural polymers.					
L4	Display a detaile	ed knowledge of the use of spectroscopy in substance identification					
L5	Describe commo	on organic chemistry laboratory procedures					
Emplo	oyability Skills	and Personal Development Planning (PDP) Skills					
SCQF	Headings	During completion of this module, there will be an opportunity to achieve core skills in:					
	ledge and standing (K )	SCQF Level <b>9</b> Students should demonstrate a broad and integrated knowledge of general organic mechanisms, carbanions and their applications in organic synthesis, the importance of protecting and blocking groups, and the chemistry of selected heterocyclic molecules. They should also demonstrate a critical knowledge of the underlying principals and concepts behind these topics.					
Practice: Applied Knowledge and Understanding		SCQF Level 9 Students should be able to describe a selection of principal skills and practices in the chemical laboratory in order to carry out a series of laboratory investigations					
Gener skills	ric Cognitive	SCQF Level Choose an item. Use the concepts and information provided to analyse problems in organic synthesis					

Communication, ICT and Numeracy	SCQF Level 9			
Skills	Students should be able to use a range of IT skills to retrieve and present in written form information from scientific data bases to support their studies. Students should be able to demonstrate the use of specialist Chemical software.			
Autonomy, Accountability and	SCQF Level 7			
Working with others	Students exercise autonomy and initiative in using IT skills in the production of coursework			
Pre-requisites:	Before undertaking this module the student should have undertaken the following:			
	Module Code: Module Title: Organic Chemistr   Chem08002 Module Title: Organic Chemistr			
	Other: Or equivalent			
Co-requisites	Module Code:	Module Title:		

\*Indicates that module descriptor is not published.

Learning and Teaching							
In line with current learning and teaching principles, a 20-credit module includes 200 learning hours, normally including a minimum of 36 contact hours and maximum of 48 contact hours.							
<b>Learning Activities</b> During completion of this module, the learning activities undertaken to achieve the module learning outcomes are stated below:	Student Learning Hours (Normally totalling 200 hours): (Note: Learning hours include both contact hours and hours spent on other learning activities)						
Lecture/Core Content Delivery	12						
Tutorial/Synchronous Support Activity	12						
Laboratory/Practical Demonstration/Workshop	24						
Hours Total							
**Indicative Resources: (eg. Core text, journals, inter	met access)						

# The following materials form essential underpinning for the module content and ultimately for the learning outcomes:

Organic Chemistry 9th Edition, John McMurray (2015), Brooks/Cole ISBN-13:978-1305080485 An Introduction to Drug Synthesis, G. L. Patrick (2015), Oxford University Press, ISBN-978-019-870843-8

(\*\*N.B. Although reading lists should include current publications, students are advised (particularly for material marked with an asterisk\*) to wait until the start of session for confirmation of the most up-to-date material)

### Attendance and Engagement Requirements

In line with the <u>Student Attendance and Engagement Procedure</u>: Students are academically engaged if they are regularly attending and participating in timetabled on-campus and online teaching sessions, asynchronous online learning activities, course-related learning resources, and complete assessments and submit these on time.

For the purposes of this module, academic engagement equates to the following:

Students are expected to attend all classes. Submit coursework and engage regularly with the VLE.

# **Equality and Diversity**

The University's Equality, Diversity and Human Rights Procedure can be accessed at the following link: <u>UWS Equality</u>, <u>Diversity and Human Rights Code</u>.

Please ensure any specific requirements are detailed in this section. Module Coordinators should consider the accessibility of their module for groups with protected characteristics.

(N.B. Every effort will be made by the University to accommodate any equality and diversity issues brought to the attention of the School)

Supplemental Information

Divisional Programme Board	Physical Sciences
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Assessment Results (Pass/Fail)	Yes □No ⊠
School Assessment Board	Physical Sciences
Moderator	Dr Callum McHugh
External Examiner	M Symes
Accreditation Details	This module is accredited by the Royal Society of Chemistry (RSC) as part of the BSc (Hons) Chemistry Programme.
Changes/Version Number	2.19 Module summary updated Module Delivery: From Hybrid-C to Face-to-Face Assessment: Change from "unseen open book" to Class test. Indicative Resources: Updated

### Assessment: (also refer to Assessment Outcomes Grids below)

Assessment is based on the following: class test (unseen) worth 50% of the mark and coursework assessment worth 50% of the mark

The continuous assessment component in this module will consist of the following elements: (i) class tests worth 20% of the final mark, and (ii) laboratory reports worth 30% of the final mark.

(N.B. (i) **Assessment Outcomes Grids** for the module (one for each component) can be found below which clearly demonstrate how the learning outcomes of the module will be assessed.

(ii) An **indicative schedule** listing approximate times within the academic calendar when assessment is likely to feature will be provided within the Student Module Handbook.)

# Assessment Outcome Grids (See Guidance Note)

Component 1								
Assessme nt Type (Footnote B.)	Learning Outcome (1)	-	Learning Outcome (3)	-	Learning Outcome (5)	Weighting (%) of Assessment Element	Timetable d Contact Hours	
Unseen closed book exam	~	~	✓			50	4	

Component 2								
Assessme nt Type (Footnote B.)	Learning Outcome (1)	-	Learning Outcome (3)	Learning Outcome (4)	Learning Outcome (5)	Weighting (%) of Assessment Element	Timetable d Contact Hours	
Class Test	~	~	~	~		20	2	
Report of Practical Work				~	~	30	0	
	Combined Total for All Components					100%	4 hours	